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Publication number: **0 432 262 A1**

(12)

**EUROPEAN PATENT APPLICATION**  
published in accordance with Art.  
158(3) EPC

(21) Application number: 88903366.8

(51) Int. Cl.<sup>5</sup> G01K 11/20

(22) Date of filing: 08.04.88

(86) International application number:  
PCT/JP88/00361

(87) International publication number:  
WO 88/08123 (20.10.88 88/23)

(30) Priority: 09.04.87 JP-87731/87

(43) Date of publication of application:  
19.06.91 Bulletin 91/25

(84) Designated Contracting States:  
BE DE FR GB IT NL

(71) Applicant: TERUMO KABUSHIKI KAISHA  
No. 44-1, Hatagaya 2-chome, Shibuya-ku  
Tokyo 151(JP)

(72) Inventor: NAKAMURA, Hideki  
Terumo Kabushiki Kaisha 2656-1, Ohbuchi  
Fuji-shi Shizuoka 417(JP)

(74) Representative: Gillard, Marie-Louise et al  
Cabinet Beau de Loménie 55, Rue  
d'Amsterdam  
F-75008 Paris(FR)

(54) THERMOMETER.

(57) This invention discloses an optical thermometer. This thermometer is equipped with a light emitting member which emits the ray of light with an intensity which is dependent on temperature when light is applied thereto, radiation means for applying stable rays of light to said light emitting member, light receiving means for receiving the rays of light emitted from said light emitting member, conversion means for converting the intensity of emitted rays of light received by the light receiving means to a corresponding temperature value, and output means for outputting converted temperature data.

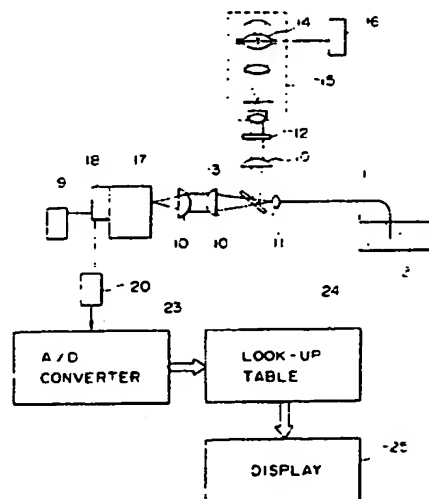


FIG. 6

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## TEMPERATURE MEASURING DEVICE

TECHNICAL FIELD

This invention relates to a temperature measuring device for measuring temperature by utilizing a light-emitting element which exhibits temperature dependence.

BACKGROUND ART

In general, methods of measuring temperature in the prior art rely upon measurement of an electric signal from a thermistor (in which a change in electrical resistance due to temperature is measured), a thermocouple (in which a thermoelectromotive force produced between different metals is measured), or the like. In any case, if the object whose temperature is being measured is a dielectric, an electrolytic solution or the like, the sensor that measures the temperature must be coated with an insulator. Consequently, there is a limit upon the extent to which the device can be miniaturized and there is a limit upon the kinds of objects that can be measured. A further disadvantage is that external noise due to electromagnetic induction tends to occur.

Methods of measuring temperature optically include a method in which the attenuation characteristic of fluorescence with respect to temperature is measured, the fluorescence being emitting by a fluorescent body comprising an inorganic compound such as a rare earth metal, and a method in which the degree of shift in a fluorescence spectrum due to temperature is detected. However, the apparatus in either case is complicated and costly.

Japanese Patent Application Laid-Open No. 58-182520 discloses a method of measuring temperature by sensing the intensity of fluorescence, but a specific technique involving the type of fluorescent body or the temperature characteristic is not indicated and no solution to the problems of the prior art is given. Moreover, since the relation between temperature and light-emitting intensity is not linear, a large error is a problem.

DISCLOSURE OF THE INVENTION

The present invention has been devised in view of the aforementioned prior art and seeks to provide a temperature sensor which is highly safe, highly reliable and free of cost-related problems.

Specifically, the temperature measuring device of the present invention comprises a light-emitting element for emitting light of an intensity dependent upon temperature when irradiated with light, irradiating means for irradiating the light-emitting element with stabilized light, light-receiving means for receiving light emitted by the irradiated light-emitting element, converting means for making a conversion into a temperature value on the basis of the intensity of the emitted light received by the light-receiving means, and output means for outputting temperature information resulting from the conversion.

Further, in accordance with one embodiment of the invention, the light-emitting element preferably is a polypyridine metal complex, especially tris(2,2'-bipyridine) ruthenium, by which temperature measurement is made possible over a wide range and with good precision.

Further, in accordance with an embodiment of the invention, the light produced by the irradiating means preferably possesses at least a wavelength which excites the light-emitting element. This will make it possible for the light-emitting element to be made to emit light highly efficiently.

Further, in accordance with an embodiment of the invention, the light-receiving means preferably includes spectral diffraction means for passing only the light emitted by the light-emitting element, whereby it is possible to eliminate the irradiating light produced by the irradiating means.

Further, in accordance with an embodiment of the invention, the light received by the light-receiving means has a wavelength in the vicinity of the peak position of the light emitted by the light-emitting element. This makes it possible to receive light having an intensity of a certain magnitude.

Further, according to an embodiment of the invention, it is preferred that an optical fiber be used as a medium through which the light-emitting element is irradiated by the irradiating means. This makes it possible to irradiate the light-emitting element with light from a remote location.

In this case, the light-emitting element is fixed to an end of the optical fiber, thereby making it possible to irradiate the light-emitting element in a sure manner.

Further, according to an embodiment of the invention, it is preferred that an optical fiber be used as a medium through which the light emitted by the light-emitting element is received by the light-receiving means. This makes it possible to receive light from a remote light-emitting element.

In this case, it is preferred that the light-emitting element be fixed to an end of the optical fiber, thereby making it possible to receive the emitted light from the light-emitting element in a sure manner.

Further, according to an embodiment of the

invention, the light-emitting element is fixed to one end of the optical fiber, the irradiating means irradiates the light-emitting element via the optical fiber, and the light-receiving means receives the emitted light via the optical fiber.

Further, according to an embodiment of the invention, the converting means preferably makes the conversion into a temperature value based on a look-up table. This makes it possible to effect a conversion into temperature information through a simple arrangement.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figs. 1 through 4 are views illustrating various method of fixing a polypyridine metal complex to an optical fiber;

Fig. 5 is a block diagram of an apparatus for measuring the intensity of emitted light from a light-emitting element used in an embodiment;

Fig. 6 is a block diagram of a temperature measuring device in the embodiment;

Fig. 7 is a graph showing the relation between temperature and emitted light intensity according to the embodiment; and

Fig. 8 is a graph showing the relation between temperature and emitted light intensity according to another embodiment.

#### BEST MODE FOR CARRYING OUT THE INVENTION

Embodiments of the present invention will now be described in detail with reference to the accompanying drawings.

The principle of the present invention will be described first.

By way of example, a polypyridine ruthenium (II) complex generally has an absorption spectrum near a wavelength of 450 nm and is excited by light having a wavelength in this vicinity (or lower). When the polypyridine ruthenium (II) complex returns from the excited state to a ground state, it emits light, at a certain light-emitting efficiency, having a peak in the vicinity of a wavelength of 600 nm. (Strictly speaking, phosphorescence and not fluorescence occupies the greater portion of this light.) Moreover, the inventor has confirmed that the light-emitting efficiency varies linearly with a change in temperature. In other words, the intensity of the emitted light is proportional to the light-emitting efficiency if the intensity of the excitation light is the same. This means that temperature can be sensed by measuring the intensity of the emitted light.

By contrast, organic fluorescent substances, e.g., many so-called polynuclear aromatic compounds such as pyrene, anthracene or fluorescein, exhibit a decrease in light-emitting efficiency due

to temperature. However, this tendency diminishes sharply at a certain temperature so that these substances are not practical for use in measuring temperature. An inorganic fluorescent body comprising a rare earth metal compound is high in cost, requires special care in handling and in unsuitable for use at room temperatures.

In the temperature measuring device of the present invention, the abovementioned polypyridine metal complex is used for the portion that senses temperature, and the metal complex is fixed to the distal end of an optical fiber. This makes possible remote measurement of very small areas and also makes it possible to minimize optical loss due to reflection and diffraction of the excitation light and emitted light at the interface between different types of substances. However, depending upon the form of the region measured, the optical fiber is not always required and the invention is not limited to an apparatus employing the optical fiber.

The polypyridine metal complex mentioned herein is the general term for a metal complex in which a polypyridine such as 2,2'-bipyridine or 1,10-phenanthroline and their derivatives are the ligands. The metal can be selected from the group of transition metals such as ruthenium, osmium, chrome, iridium, iron, cobalt and europium. In particular, selecting ruthenium (II) ion as the metal ion is ideal since this will provide the highest sensitivity. Specific examples of the latter are tris(2,2'-bipyridine) ruthenium (II) complex chlorides, tris(1,10-phenanthroline) ruthenium (II) complex perchlorates, tris(4,7-diphenyl-1,10-phenanthroline) ruthenium (II) complex perchlorates, etc. Of course, any other polypyridine metal complex can be used.

Next, various methods of fixing this polypyridine metal complex to an optical fiber will be considered. Examples of the fixing requirements are as follows:

(1) If the object undergoing measurement is a liquid, the fixed metal complex should not dissolve into the liquid (dissolution will cause conversion of the emitted light intensity and make stable measurement difficult).

(2) A non-reversible change due to a chemical reaction should not arise between the device and the substance whose temperature is being sensed.

Figs. 1 through 4 illustrate various examples of fixing the polypyridine metal complex.

Fig. 1 shows an example in which a polypyridine metal complex 2 is dissolved or dispersed in a polymer 3 thereby to be fixed to the tip of an optical fiber serving as a supporting member for the metal complex. This represents the ideal fixation method.

In this case, there are many examples which can be used as the polymer 3. For instance,

general-purpose plastics which are widely available can be used. Examples are low-density polyethylene, polypropylene, polyvinyl chloride, ethylene acetate-vinyl copolymer, polystyrene, polymethyl methacrylate, silicone resin and polyurethane. It is also permissible to use plastics obtained by plasticizing these polymers with a plasticizer. Though the support member is not particularly required, it is preferred that an optical fiber be employed when the support member is used. Further, as shown in Fig. 2, it is permissible to affix a suitable protective film 4 or the like to the outer side (the side that comes into contact with the object whose temperature is to be measured) of the polymer 3. Also, as shown in Fig. 3, it is possible to adopt a capsule configuration enclosing a gel or solution 5 of the polypyridine metal complex, in which case the gel, aqueous solution, etc., of the polypyridine metal complex is sealed in by a suitable film 6 and fixed to the end face of an optical fiber. Here a gel refers to a semisolid and includes an aqueous polymer such as gelatin, polyacrylamide and polyacrylic sodium.

A third method is to chemically or physically adsorb the polypyridine metal complex on an adsorbing body 7, as shown in Fig. 4. Examples of the adsorbing body that can be mentioned include inorganic substances such as silica gel and glass, porous polymers, various ion-exchange resins, and natural substances such as many sugars and proteins. Though any of these adsorbing bodies can be used, employing an anion-exchange resin or chelate adsorbent makes it possible to achieve more stable fixation.

Thus, by way of example, a polypyridine metal complex is fixed to one end of an optical fiber, the other end of the optical fiber is irradiated with light (excitation light) 8 obtained by spectrally refracting light from a suitable source (e.g., a xenon lamp or ultra high-pressure mercury lamp) into an excitation spectral wavelength region (ordinarily less than 500 nm), and guiding this excitation light through the optical fiber to the portion at which the polypyridine metal complex is fixed. When this is done, the polypyridine metal complex emits reddish orange light 9 the intensity of which is commensurate with temperature. The emitted light propagates through the interior of the optical fiber and has its excitation light component separated by a suitable spectroscopic device such as an optical filter or monochromator. The light is then converted into an electric signal by a light-receiving element (a photomultiplier tube or the like) so that the intensity of the light can be measured. It should be noted that the excitation light in this case can be light which is continuous in time or light which is pulsating. Also, the optical fiber which guides the emitted light need not be the same as that which guides the excitation light.

It has been found that the temperature measuring device of the embodiment based on the foregoing principle exhibits excellent reproducibility and is capable of performing highly accurate measurement even in the vicinity of room temperature.

Specific examples of the temperature measuring device of the embodiment will now be described.

#### <Example 1>

Dissolved in 20 ml of dimethyl formamide (a reagent manufactured by Kanto Chemical Co., Inc.) were 0.01g of tris(2,2'-bipyridine) ruthenium (II) chloride (a reagent manufactured by Aldrich Chemical Co., Inc.) and 1g of polymethyl methacrylate (a reagent manufactured by Aldrich Chemical Co., Inc., having a molecular weight of 12,000). One end of a 2 m length of plastic optical fiber (SK-10, manufactured by Mitsubishi Rayon Co., Ltd., having an outer diameter of 0.25 mm) was dipped in the resulting solution, removed after about 10 sec and then dried under reduced pressure. This made it possible to fix the tris(2,2'-bipyridine) ruthenium (II) chloride to the tip of the optical fiber using the polyethyl methacrylate. Thus, a temperature probe having the form shown in Fig. 1 was manufactured. An apparatus of the kind shown in Fig. 5 was constructed using the temperature probe thus produced and equipment constituting an optical system. The graphs have a vertical axis along which the relative intensity of emitted light is plotted, with the relative intensity of received light at 25°C being taken as "10".

In Fig. 5, numeral 1 denotes the optical fiber having the above-described temperature probe formed on its distal end, which is immersed in water in a isothermal water bath 21. Numeral 22 denotes a mercury thermometer for measuring the temperature of the water in the isothermal water bath 21.

When a ultra high-pressure mercury lamp 14 in a parallel luminous flux irradiating device 15 is ignited by a starter 16, the light passes through an interference filter 12 and a plano-convex lens 10 and is reflected by a dichroic mirror 13. The reflected light passes through an objective lens 11 for a microscope. As a result, the temperature probe situated at the terminal of the optical fiber 1 is irradiated with excitation light having a center wavelength of 435 nm. In response to this irradiation, the temperature probe [tris(2,2'-bipyridine) ruthenium (II) chloride] is caused to emit light. The emitted light passes through the objective lens 11 and dichroic mirror 13 and is condensed on a monochromator 17 by two plano-convex lenses 10. Only monochromatic light having a wavelength of 620 nm is transmitted by the monochromator 17.

This light is converted into an electric signal by a photomultiplier tube 18. The electric signal is amplified by an amplifier 20, thereby making it possible to measure the intensity of the 620 nm wavelength light emitted by the [tris(2,2'-bipyridine) ruthenium (II) chloride]. Numeral 19 denotes a power supply for the photoelectron multiplier tube 18.

The relationship between the light intensity measured by this apparatus and the temperature of the water in the isothermal water bath 21 (namely the value indicated by the mercury thermometer 22) is as shown in Fig. 7. This shows that an excellent linear relationship with respect to temperature was obtained. Accordingly, this makes it possible to subsequently measure the temperature of an object from this relationship between temperature and light intensity.

The three plano-convex lenses 10 in Fig. 5 were model numbers 01LPX277 manufactured by Melles Griot, the objective lens 11 for the microscope was manufactured by Olympus K.K., the interference filter 11 was model number BPF-4 (having a center wavelength of 435 nm) manufactured by Vacuum Optics Corp. of Japan, and the dichroic mirror 13 was a blue reflective mirror manufactured by Vacuum Optics Corp. of Japan. A ultra high-pressure mercury lamp USH-102D manufactured by Ushio Inc. was used as the light source, and model numbers UI-100Q, HB-10102AA, both manufactured by Ushio Inc., were used as the parallel luminous flux irradiating device 15 and ultra high-frequency mercury lamp starter 16, respectively. A model number H-20V manufactured by Jobin Yvon was used as the monochrometer 17, and model numbers R1477, C665, both manufactured by Hamamatsu Photonics K. K., were used as the photomultiplier tube 18 and high-voltage power supply 19 for this photomultiplier tube, respectively. A digital electrometer FC7401, manufactured by Iwatsu Electric Co. Ltd. was used as the amplifier 20.

An apparatus of the kind shown in Fig. 6 was constructed in order to actually measure temperature using this probe.

A description of the elements at numbers 1 through 20 in Fig. 6 is omitted in order to avoid redundancy.

In Fig. 6, numeral 23 denotes an A/D converter for converting an analog signal from the amplifier 20 into digital data, and numeral 24 denotes a look-up table for outputting data corresponding to the digital data.

The look-up table 24 comprises a ROM to which the data from the A/D converter 23 is inputted as an address so that the ROM may output the corresponding data. It goes without saying that the relationship between the inputted address and the outputted data is the relationship shown in Fig. 7.

The digital data (temperature information) resulting from the conversion performed by the look-up table 24 is latched in an LCD driver of a display unit 25 and displayed.

Though the description is out of sequence, a linear equation stored in the look-up table is determined by the method of least squares.

#### <Example 2>

The present invention is not limited to the temperature probe described above. For example, one end of an optical fiber the same as that used in Example 1 was dipped into a 5% alcohol solution (No. 27,470-4, a reagent manufactured by Aldrich Chemical Co., Inc.) of Nafion 107, which is a cation-exchange resin, for about 10 sec. followed by drying. Only a length of about 2 mm of the tip of the optical fiber was thus coated with the cation-exchange resin. The coated portion was then dipped in a 20 mM aqueous solution of tris(2,2'-bipyridine) ruthenium (II) chloride for 1 min. followed by washing with water.

An excellent linear relationship of the kind shown in Fig. 8 was obtained when temperature and the intensity of emitted light were determined in the same manner as in Example 1 using the probe (which corresponds to Fig. 2) having the tris(2,2'-bipyridine) ruthenium (II) cation fixed thereto.

In accordance with the present embodiment of the invention as described above, there can be obtained a temperature sensor which is highly safe and reliable and free of cost-related problems. In particular, since a relationship is obtained in which the intensity of emitted light in the room-temperature region maintains a linear state, the sensor is ideal for measuring body temperature or the temperature of blood at any in vivo location.

By using the optical fiber as a medium for optically irradiating the polypyridine metal complex serving as the light-emitting element and as a medium for inputting the emitted light, it becomes possible to perform measurement accurately and remotely even in locations where there is a great amount of electromagnetically induced noise.

Further, by using a single optical fiber for optical irradiation and for inputting emitted light, it becomes possible to measure temperature with facility even when the object to be measured is concealed.

In the present embodiment, it is described that a mercury lamp is used as a light source for irradiating the polypyridine metal complex. However, since a light source having a sufficient optical intensity even below a wavelength of about 500 nm is advantageous, it is permissible to use a xenon lamp, a tungsten lamp or the like. In addition, though a photomultiplier tube is used in the present

embodiment as means for sensing temperature, it is permissible to use a photodiode or the like. Accordingly, the present invention should be interpreted in accordance with the claims thereof and the above-described embodiment can be modified and altered without departing from the scope of the claims.

## Claims

1. A temperature measuring device comprising:  
a light-emitting element for emitting light of an intensity dependent upon temperature when irradiated with light;  
irradiating means for irradiating said light-emitting element with stabilized light;  
light-receiving means for receiving light emitted by the irradiated light-emitting element;  
converting means for making a conversion into a temperature value on the basis of the intensity of the emitted light received by said light-receiving means, and  
output means for outputting temperature information resulting from the conversion.
2. A temperature measuring device according to claim 1, characterized in that said light-emitting element is a polypyridine metal complex.
3. A temperature measuring device according to claim 1, characterized in that a tris(2,2'-bipyridine) ruthenium (II) chloride is used as the polypyridine metal complex.
4. A temperature measuring device according to claim 1, characterized in that the light produced by said irradiating means preferably at least possesses wavelength which excites said light-emitting element.
5. A temperature measuring device according to claim 1, characterized in that said light-receiving means includes spectral diffraction means for passing only the light emitting by said light-emitting element.
6. A temperature measuring device according to claim 1, characterized in that the light received by said light-receiving means is a wavelength spectrum in the vicinity of the peak position of the light emitting by said light-emitting element.
7. A temperature measuring device according to claim 1, characterized in that an optical fiber is used as a medium through which said light-emitting element is irradiated by said irradiat-

ing means.

8. A temperature measuring device according to claim 7, characterized in that said light-emitting element is fixed to an end of said optical fiber.
9. A temperature measuring device according to claim 1, characterized in that an optical fiber is used as a medium through which the light emitted by said light-emitting element is received by said light-receiving means.
10. A temperature measuring device according to claim 9, characterized in that said light-emitting element is fixed to an end of the optical fiber.
11. A temperature measuring device according to claim 1, characterized in that said light-emitting element is fixed to one end of an optical fiber, said irradiating means irradiates said light-emitting element via said optical fiber, and said light-receiving means receives the emitted light via said optical fiber.
12. A temperature measuring device according to claim 1, characterized in that said converting means makes the conversion into a temperature value based on a look-up table.

FIG. 1

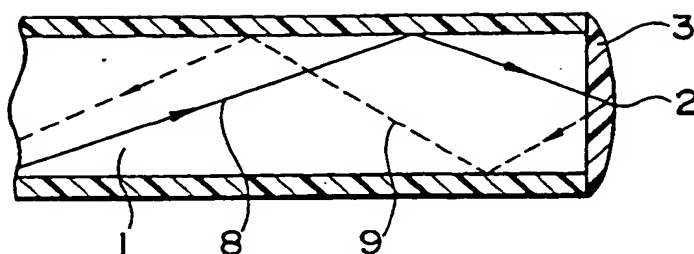


FIG. 2

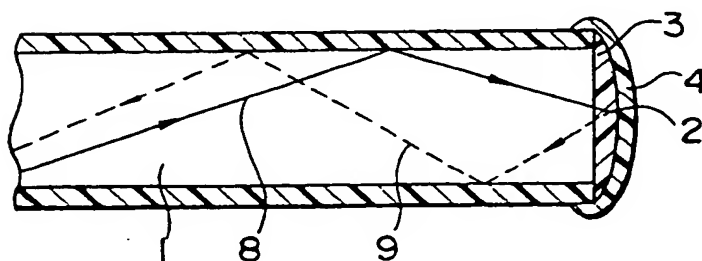


FIG. 3

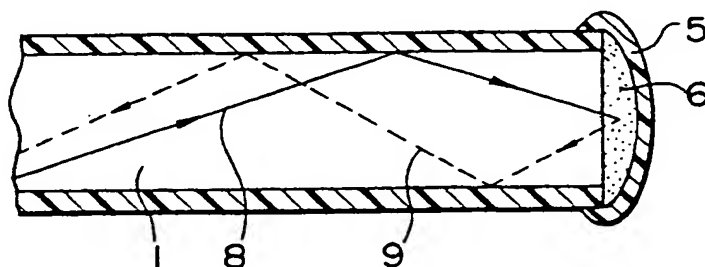
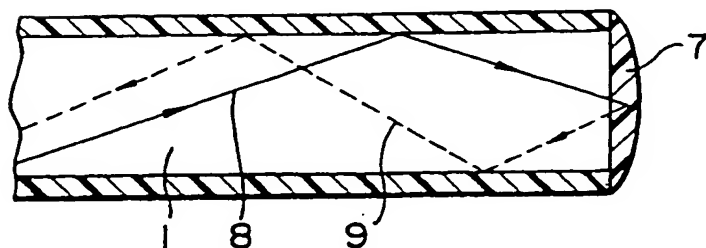
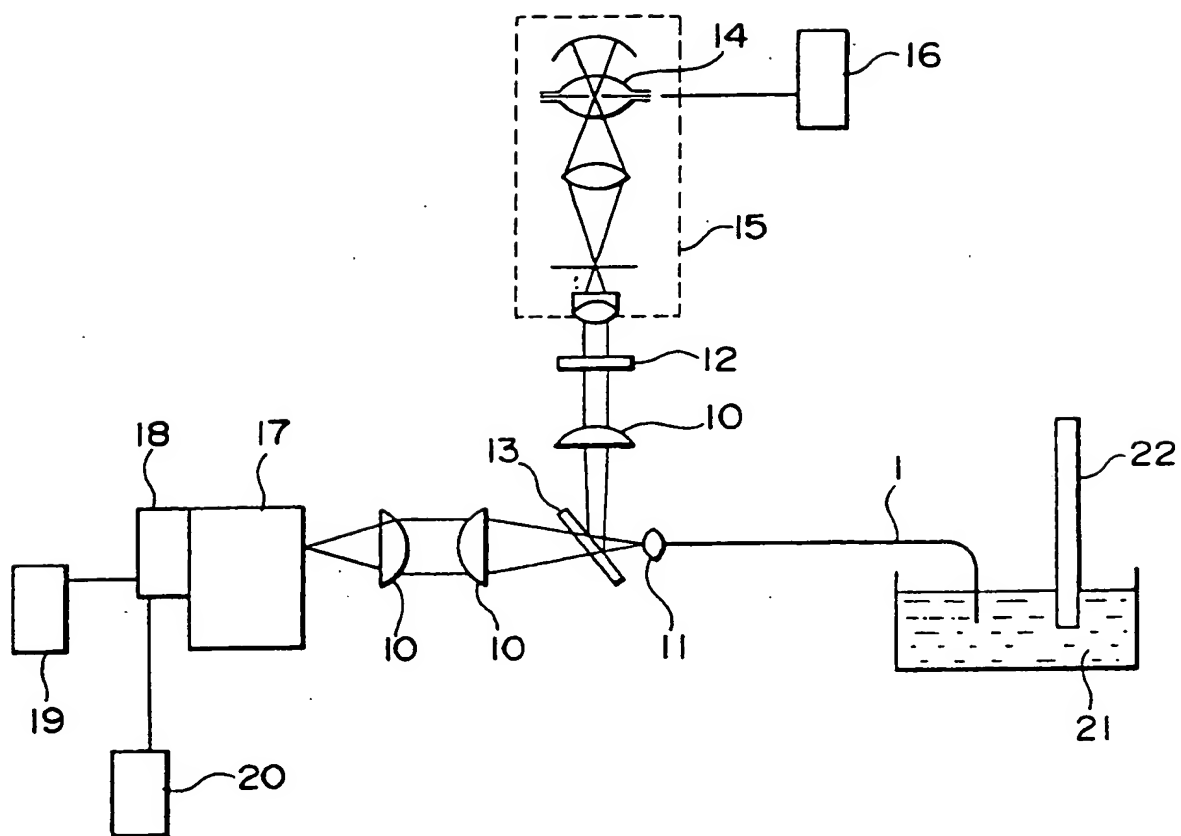


FIG. 4





F I G. 5



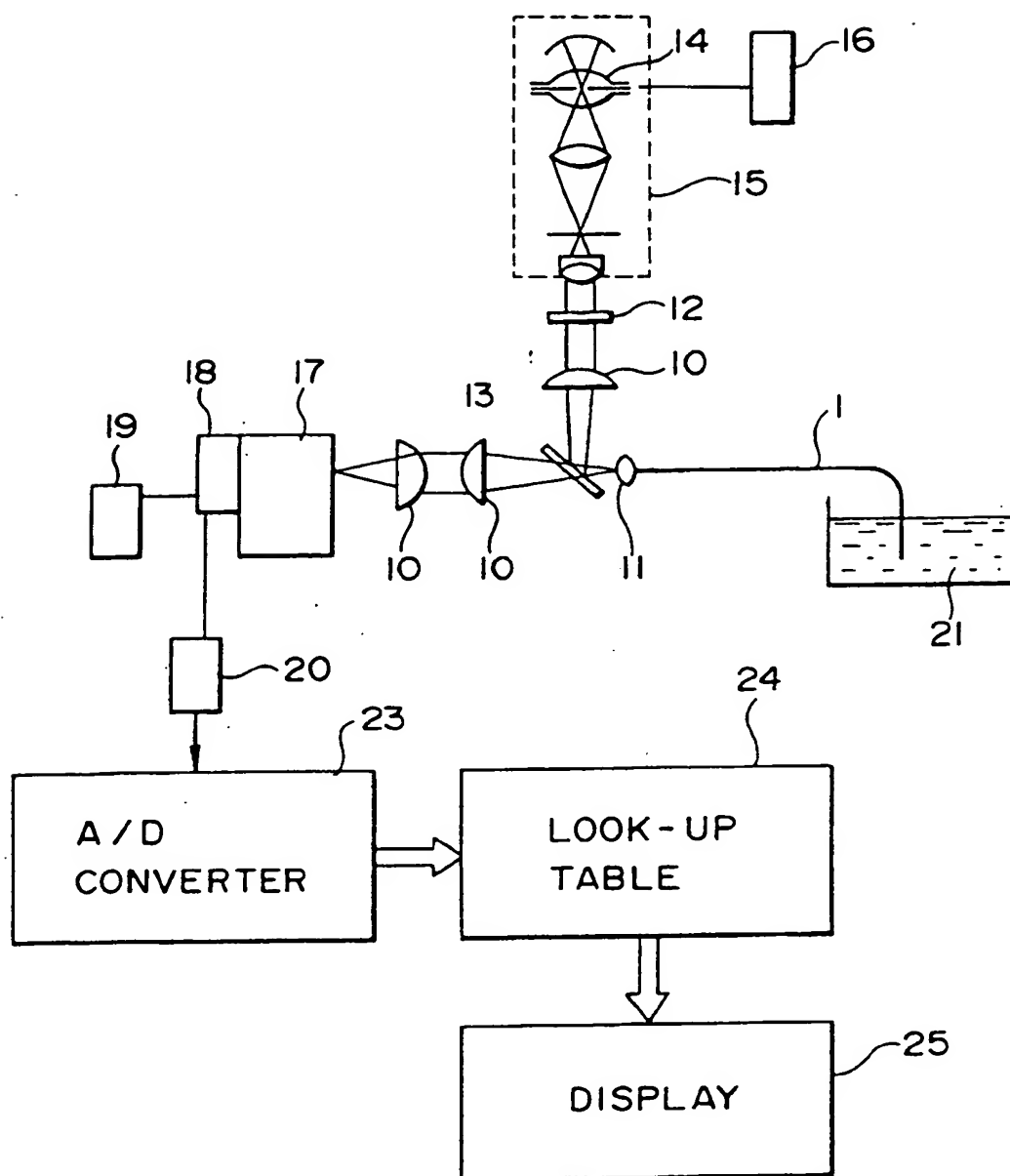
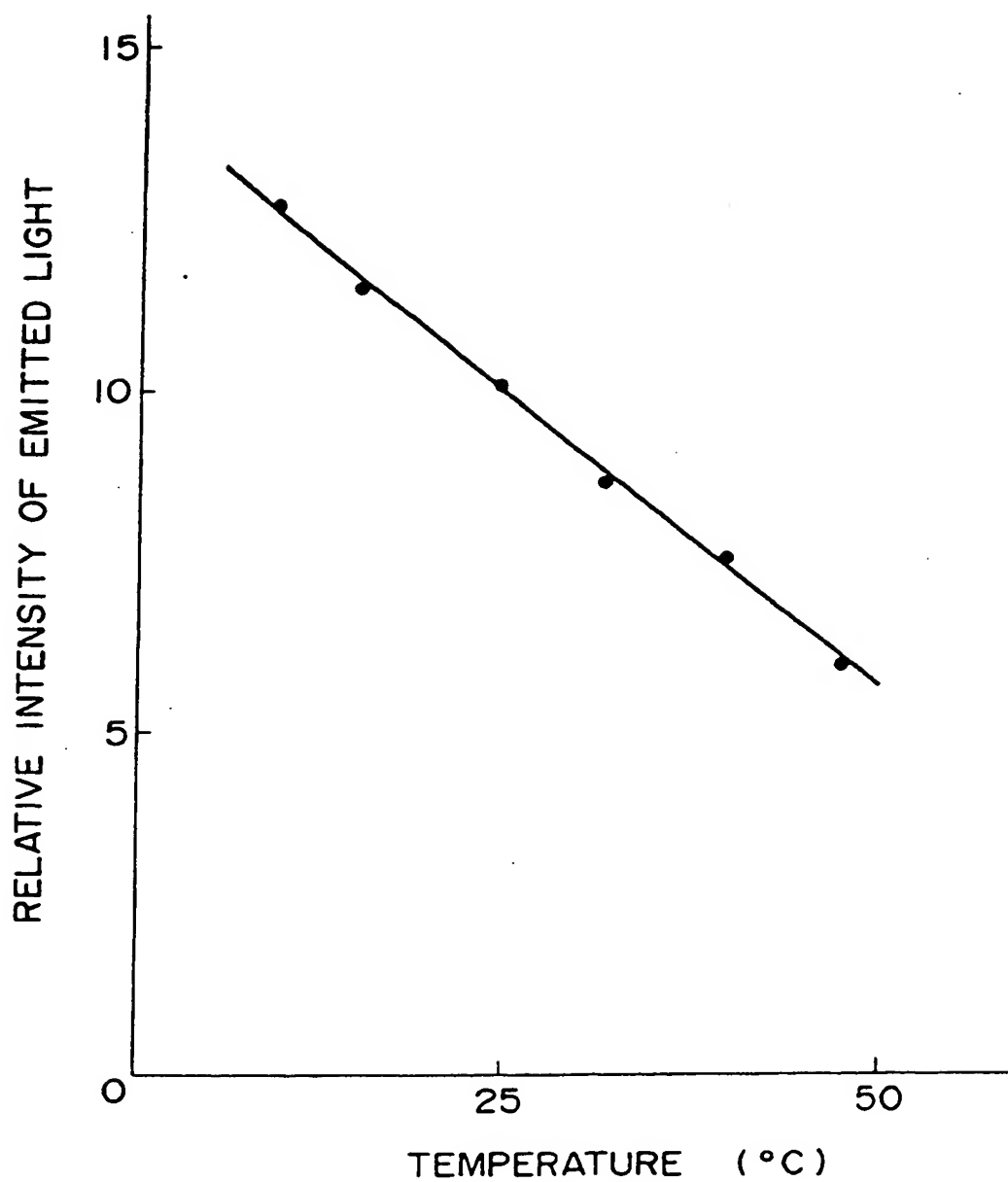


FIG. 6



F I G. 7

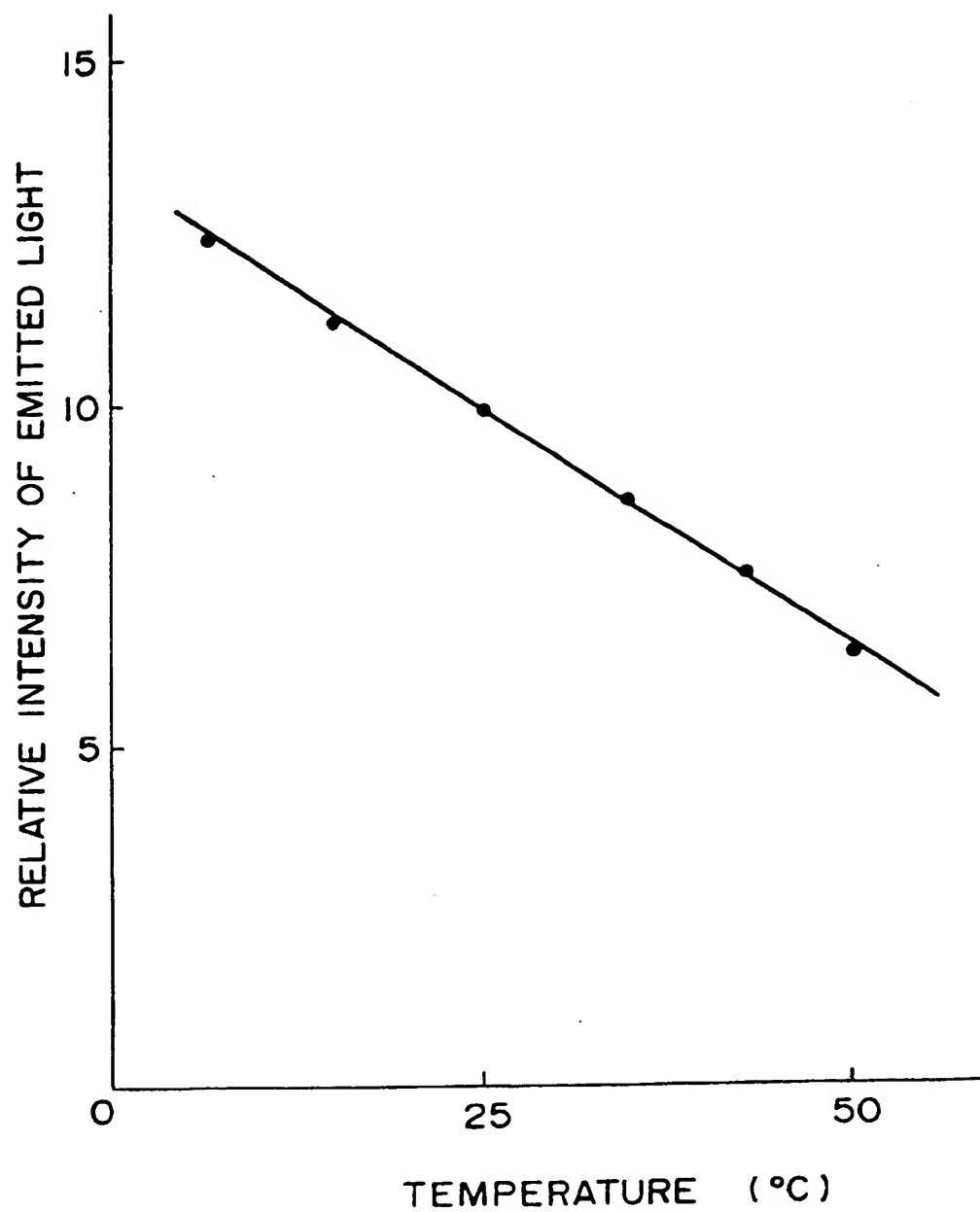


FIG. 8

# INTERNATIONAL SEARCH REPORT

International Application No PCT/JP88/00361

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) *		
According to International Patent Classification (IPC) or to both National Classification and IPC		
Int.Cl <sup>4</sup> G01K11/20		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched *		
Classification System 1	Classification Symbols	
IPC	G01K11/12, 11/20	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched *		
Kokai Jitsuyo Shinan Koho	1971 - 1987	
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT *</b>		
Category *	Citation of Document, ** with indication, where appropriate, of the relevant passages **	Relevant to Claim No. **
X	JP, A, 53-116882 (Toshiba Corp.) 12 October 1978 (12. 10. 78) (Family: none)	1, 4-11
X	JP, A, 58-30628 (Machida Opto Giken Yugen Kaisha) 23 February 1983 (23. 02. 83) (Family: none)	1, 4-12
<p>* Special categories of cited documents: **</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents such combination being obvious to a person skilled in the art</p> <p>"S" document member of the same patent family</p>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search	Date of Mailing of this International Search Report	
June 16, 1988 (16. 06. 88)	June 27, 1988 (27. 06. 88)	
International Searching Authority	Signature of Authorized Officer	
Japanese Patent Office		

Form PCT ISA-210 (second sheet) (January 1985)